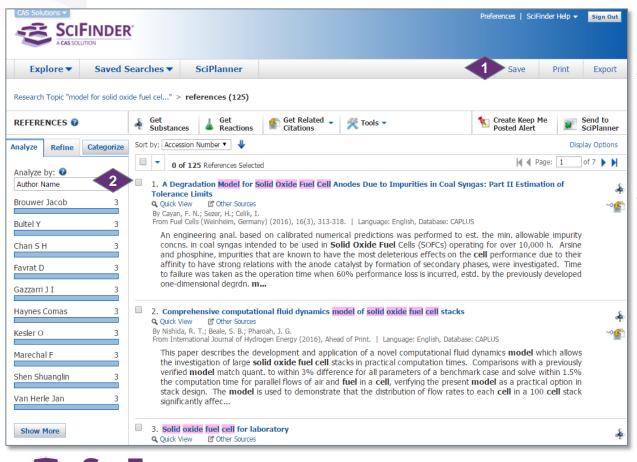
## How to... Save, Print and Export Answers

### Keep your SciFinder® answers for future use

Keep answer sets for future use with print, save and export capabilities. To generate a hardcopy of part or all of your answer set, use Print to create a .pdf file that can be viewed, saved and printed. Save your answers to the SciFinder server for future use or Export answers to your computer to use with other software applications or to share with colleagues. To access additional training resources on this and other topics, consult the online Help or visit www.cas.org/training/scifinder.

### Save, Print and Export: General Information



You can **Save**, **Print** or **Export** reference, substance and reaction answer sets.

- In the upper right, click **Save**, **Print** or **Export** to launch a dialog window.
  - The options in the dialog window will vary depending on the type of answer set and whether you are saving, printing or exporting.

By default, all answers are saved, printed or exported.

- An option to save, print or export only selected answers is available.
- To select an answer, click the box to the left of an answer number.

The following pages show examples of printing, saving and exporting answer sets.



CAS is a division of the American Chemical Society. Copyright 2017 American Chemical Society. All rights reserved.

### **Save Reactions Example**

Explore <b>v</b>	Saved S	Searches 🔻	SciPlanner			Save	Print	Export
Reaction Structure stru	ucture variab	le only at spe	> reactions (1821	)				
REACTIONS 😧		Get Reference	es 🕺 Tools 🗸				💓 S	Send to SciPlanner
Analyze Refine		Group by: No	Grouping <ul> <li>Grouping</li> <li>Sort</li> </ul>	by: Relevance	• 4		Displ	lay Options
Analyze by: 🔞		· · 0	of 1821 Reactions Sele	ected		┥ 🌗 Pag	je: 1 of	f 92 🕨 射
Reagent	¥	1. View	Reaction Detail	🕫 Link				
NaHCO3	1055	Single S	tep Hover over any	structure for m	ore options.			
NaOH	757							
HCI	716				CH₃ I			
H <sub>2</sub> SO <sub>4</sub>	574	CHs		0 0	çı 🔨			
H <sub>2</sub> O	501	L)	С • он-	Sон	$\rightarrow$			
Et <sub>3</sub> N	401			II O				
КОН	279		~122 찬					
K <sub>2</sub> CO <sub>3</sub>	229				100% ~74			
								_
AcCl	140	+ Ov	erview	2	Save This Answer Set			
716-61-0	120	► Exp	erimental Proced	ure				* Require
Show More			Reaction Detail	a Link u ciwi	Save:			
		- 2. <b>VIEW</b>	Reaction Detail of	Silli 🔊 Silli	<ul> <li>All answers</li> <li>Only selected answers</li> </ul>			
					Title: *			
					Clopidogrel reactions			
					Description:			
					One and two step reactions			-
								-
						3	► OK	Cancel

Use **Save** to place a file of your reference, substance or reaction answers on the SciFinder server. The answers are saved with your SciFinder login ID so that you can access them from any computer.



In the upper right, click Save.

In the **Save This Answer Set** window, select the answers to save, enter a **Title** and an optional **Description**.



Click **OK** to save the answers.

The **Save This Answer Set** dialog window closes and you are returned to your active session. To place answers on your own computer or network, use **Export**.

_	
	n
	μ

For a single answer set, you can save up to 20,000 answers.



CAS is a division of the American Chemical Society. Copyright 2017 American Chemical Society. All rights reserved.

# Work with Saved Answer Sets

Saved Searches -Saved Answer Sets Keep Me Posted History

Click the drop-down arrow on the Saved Searches tab to access all of your Saved Answer Sets, Keep Me Posted alert results and your search History.

		Answer Sets	2			
SAVED SEARCHES		2 of 14 Substa	ance Answer Sets Selected References (30)	Substances (14)	Reactions (12)	
Saved Answer Sets		Subs HTG 2016 5	5 6 (535)	Edit	GO Link	Saved May 6, 201
Keep Me Posted History		Chemical Structu (535)	re substructure > substances			
		Subs assoc'd with (2057)	n flavors in mouthwash; rxn availa	Edit	\varTheta Link	Saved Nov 10, 201
		references (688)	flavoring for mouthwash" > > keep analysis "CA Section Title" tances (3145) > keep analysis bility" (2057)			
		chemsearcher test (380)		Edit	\varTheta Link	Saved Sep 8, 20
		Property "Experir 18.2" > substanc	nental - Optical Rotatory Power, es (380)			
		coord cmpds 25 (25)		Edit	👄 Link	Saved Aug 18, 201
			re substructure > substances atom Attachment" (25)			
		coord cmpd exar	nple (3262)	Edit	\varTheta Link	Saved Aug 18, 20
<u>م</u>		using real atom a	attachment			_
4 Co	ombine	Answer Sets				
S	elect an	option for combi	ning the two selected saved answer se	ts:		
	Q	Combine	Include all substances from both se	ts		
		Intersect	Include only substances that appea	ir in both sets		
	Q	) Exclude	Include only answers from coord cr that are not in coord cmpd example			
	0	Exclude	Include only answers from coord cr	npd example		

Include only answers from coord cmpd example that are not in coord cmpds 25

> Combine Answer Sets Cancel

On the top navigation bar, click Saved Answer Sets to open the Saved Answer Sets dialog window.

Saved reference, substance and reaction answer sets are available on separate tabs.

You can click:

- A title to re-open the answer set
- Edit to make changes to the title or description
- Link to create a link that you can bookmark for quick access or send to colleagues who can open the answer set in SciFinder
- A checkbox and the select drop down arrow to Delete Selected



If you select two or more answer sets, then the Combine Answer Sets button becomes active. Click it to open the corresponding window.

You can combine two answer sets in one of four ways. Select the option of interest.

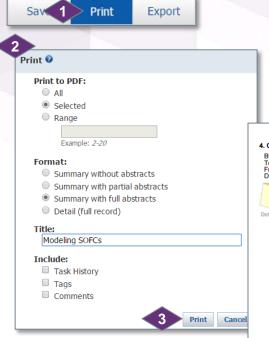
If you select more than two answer sets, then only the Combine and Intersect options are available.

Click the Combine Answer Sets button. The answers are merged and become your active session.



CAS is a division of the American Chemical Society. Copyright 2017 American Chemical Society. All rights reserved.

## **Print References Example**

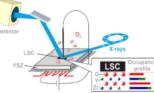


### The PDF document

#### SciFinder®

#### 4. Operando X-ray Investigation of Electrode/Electrolyte Interfaces in Model Solid Oxide Fuel Cells

By Volkov, Sergey, Vonk, Vedran; Khorshidi, Navid; Franz, Dirk; Kubicek, Markus; Kilic, Volkan; Felici, Roberto; Huber, Tobias M.: Navickas, Edvinas; Rupp, Ghislain M.; et al From Chemistry of Materials (2016), Ahead of Print, Language: English, Database: CAPLUS, DOI:10.1021/acs.chemmater.6b00351



We employed operando anomalous surface X-ray diffraction to investigate the buried interface between the cathode and the electrolyte of a model solid oxide fuel cell with at. resoln. The cell was studied under different oxygen pressures at elevated temps, and polarizations by external potential control. Making use of anomalous X-ray diffraction effects at the Y and Zr Kedges allowed us to resolve the interfacial structure and chem. compn. of a (100)-oriented, 9.5 mol % yttria-stabilized zirconia (YSZ) single crystal electrolyte below a La0.6Sr0.4CoO3.6 (LSC) electrode. We observe vttrium segregation toward the YSZ/LSC electrolyte/electrode interface under reducing conditions. Under oxidizing conditions, the interface becomes Y depleted. The yttrium segregation is corroborated by an enhanced outward relaxation of the YSZ interfacial metal ion layer. At the same time, an increase in point defect concn. in the electrolyte at the interface was obsd, as evidenced by reduced YSZ crystallog, site occupancies for the cations as well as the oxygen ions. Such changes in compn. are expected to strongly influence the oxygen ion transport through this interface which plays an important role for the performance of solid oxide fuel cells. The structure of the interface is compared to the bare YSZ(100) surface structure near the microelectrode under identical conditions and to the structure of the YSZ(100) surface prepd. under ultrahigh vacuum conditions

~0 Citinas

Copyright © 2016 American Chemical Society (ACS). All Rights Reserved

#### 10. Multiscale model for solid oxide fuel cell with electrode containing mixed conducting material

By Chen, Daifen; Wang, Hanzhi; Zhang, Shundong; Tade, Moses O.; Shao, Zongping; Chen, Huili From AIChE Journal (2015), 61(11), 3786-3803. Language: English, Database: CAPLUS, DOI:10.1002/aic.14881

Solid oxide fuel cells (SOFCs) with electrodes that contain mixed conducting materials usually show very different Some double tier cens (SOFCS) with relectiones in that contain inseed conducting initiaterials outcaing show very dimeterit relations among microstructure parameters, effective electrode characteristics, and detailed working processes from conventional ones. A new multiscale model for SOFCs using mixed conducting materials, such as LSCF or BSCF, was developed. It consisted of a generalized percolation micromodel to obtain the electrode properties from microstructure parameters and a multiphysics single cell model to relate these properties to performance details. Various constraint relations between the activation overpotential expressions and elec, boundaries for SOFC models were collected by analyzing the local electrochem. equil. Finally, taking a typical LSCF-SDC/SDC/Ni-SDC intermediate temp. SOFC as an example, the application of the multiscale model was illustrated. The accuracy of the models was verified by fitting 25 exptl. I-V curves reported in literature with a few adjustable parameters; addnl., and several conclusions were drawn from the anal. of simulation results. © 2015 American Institute of Chem. Engineers AIChE J, 2015.

~2 Citinas

Copyright © 2016 American Chemical Society (ACS). All Rights Reserved

#### 15. Solid oxide fuel cell stack based on single-sheet electrolyte

By Liu, Jiang; Zhang, Li; Liu, Yan; Yuan, Lili From Shiyong Xinxing Zhuanli Shuomingshu (2014), CN 203871424 U 20141008, Language: Chinese, Database: CAPI US



CAS is a division of the American Chemical Society. Copyright 2017 American Chemical Society. All rights reserved.

Page 1

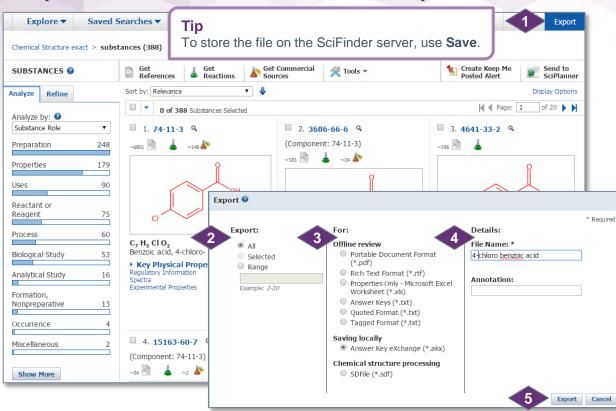
You can print references, substances and reactions. To begin, click Print in the upper right.

In the **Print** dialog window:

- 2 Select the answers you want to print: All, Selected or a Range.
  - In the Format section, click a radio button to select the parts of the record that you want to print.
  - Enter a Title.
  - If desired, specify additional information to Include with your answer set (options vary depending on the type of answer set).

Click Print to generate a .pdf file that downloads or opens in a separate window, depending on browser settings.

# **Export Substances Example**



FILE FORMAT	PURPOSE
.ris	To export references for use with citation management software (check your application to confirm which file format to use)
.akx	To export data for collaboration with other SciFinder users
.sdf	To export structures and substance identifiers to a file format readable by some molecule database programs; structures are represented in molfile format

### **CAS Customer Center**

E-mail: help@cas.org

Phone numbers: http://www.cas.org/contact-us/cas-customer-center

CAS is a division of the American Chemical Society. Copyright 2017 American Chemical Society. All rights reserved.

**Export** answers for use with other software applications or to collaborate with a colleague. When you export, the file is stored on your computer.

Click **Export** in the upper right. Specify the answers to **Export**.



Specify the file format.

- See the table for descriptions of commonly used formats.
- The file formats and options vary, based on the type of answer set you have.
- Consult the online SciFinder Help for more information about exporting data.

Under **Details**, specify a **File Name**, and, optionally, **Annotation**.

 The options in this section vary depending on the file format you select.

Click **Export**. The file is placed into your download folder or you are prompted to save it, depending on your browser settings. You are returned to your active session.

